

## Atomistic Mechanisms of Phase Stability in Mn-Based Layered Oxide Cathodes for Sodium-Ion Batteries

Manganese-based layered oxides ( $\text{Na}_x\text{MnO}_2$ ) are promising cathode materials for next-generation sodium-ion batteries (NIBs) due to their high energy density and cost-effectiveness. [1] Among the key structural phases, P2 and O3 differ in oxygen stacking, with P2 exhibiting superior sodium-ion diffusion. [1] However, its electrochemical performance is hindered by solid-state phase transitions during  $\text{Na}^+$  insertion/extraction, leading to structural degradation and capacity fade. [2] A fundamental understanding of these transitions at the atomic level is crucial for improving long-term stability. In this work, we investigate the thermodynamic and kinetic factors governing phase evolution in P2- $\text{Na}_x\text{MnO}_2$ . [3] Density functional theory (DFT) calculations reveal that phase stability and transformation pathways are dictated by cooperative Jahn–Teller effects (CJTE) from  $\text{Mn}^{3+}$  centres,  $\text{Na}^+$  ordering, and electrostatic/covalent interactions. To probe the kinetics of these transitions, we employ variable-cell nudged elastic band (VC-NEB) calculations, [4] identifying shear-driven gliding and tetrahedral transition states as dominant energy barriers in P2-to-P2' and P2-to-OP4/O2 transformations. Furthermore, we introduce structural distortion metrics as predictive descriptors of phase stability, offering a new approach to designing more resilient cathodes. [5] Our findings establish a comprehensive framework for optimizing Mn-based layered oxides by balancing thermodynamic stability and kinetic accessibility. This work provides critical design guidelines for enhancing electrochemical durability in sodium-ion batteries, advancing their viability for large-scale energy storage.

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