

# One-Pot Hydrogen Production and Carbon Valorization via Innovative Borate-boosted Bio-alcohol Conversion

Alessandra Di Nardo <sup>a,b</sup>, Gianluca Landi <sup>a</sup>, Giuseppina Luciani <sup>b</sup>, Maria Portarapillo <sup>b</sup>, Giovanna Ruoppolo <sup>a</sup>, Danilo Russo <sup>b</sup>, Almerinda Di Benedetto <sup>b</sup>

<sup>a</sup> Institute of Science and Technology for Sustainable Energy and Mobility, CNR, P. le Tecchio 80, Naples 80125, Naples, Italy

<sup>b</sup> Department of Chemical, Materials and Production Engineering, University of Naples Federico II, P. le Tecchio 80, Naples 80125, Italy

E-mail: [alessandra.dinardo@unina.it](mailto:alessandra.dinardo@unina.it)

Hydrogen is widely recognized as a pivotal energy vector to achieve net-zero emissions by 2050, serving both as a versatile energy carrier and an industrial feedstock. However, current production methods are still largely based on fossil fuels, resulting in significant CO<sub>2</sub> emissions. The development of innovative and sustainable pathways for hydrogen production is therefore critical to achieving decarbonization goals.

In this context, our research group has developed and patented (WO 2023/105545 A1) an innovative process for hydrogen production from bio-alcohols and sodium metaborate, a by-product of sodium borohydride hydrolysis. The process operates in batch reactors in series at 300 °C, feeding bio-alcohol and water alternatively to sodium metaborate. This configuration enables both hydrogen production and carbon valorization from renewable feedstocks.

Using bioethanol, the process produces high-purity hydrogen (>95% v/v) at 30-45 bar, suitable for blending into natural gas pipelines or direct use in fuel cells. Simultaneously, the process allows the production of marketable chemicals in the liquid phase and a solid mainly consisting of oligoethylene (-CH<sub>2</sub>-CH<sub>2</sub>-)<sub>n</sub>, with potential applications as synthetic lubricant and viscosity modifier [1]. Employing crude glycerol, a cheap and widely available by-product, the selectivity in hydrogen reaches 55% v/v, generating a gas stream suitable for downstream upgrading, such as methanation for synthetic natural gas (SNG) production. Additionally, the in-situ produced hydrogen enables cascading reactions leading to valuable co-products, including 1,2-propanediol and thermally stable aromatic resins in the solid phase [2].

A key innovation of the process lies in the integration of sodium metaborate, traditionally regarded as a bottleneck in hydrogen storage loop due to its regeneration issues. Here, sodium metaborate plays a dual role as both alkaline stabilizer and reaction promoter, enabling one-pot bio-alcohol dehydrogenation and oligomerization. Moreover, the use of bio-alcohols as inherently CO<sub>2</sub>-neutral reduces the carbon footprint associated with the hydrogen production chain.

This approach represents advancement in hydrogen production technologies by combining high-efficiency hydrogen generation and carbon valorization in a single pathway.

**Keywords:** *hydrogen, bio-alcohols, carbon valorisation, sodium metaborate*

## References:

<sup>1</sup> Di Nardo, A. ACS Omega. 2024, 9, 7, 7793-7805

<sup>2</sup> Di Nardo, A. Engineering Journal 2024, 500, 156634